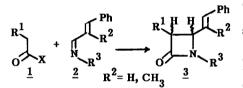
NEW STEREOCHEMICAL OUTCOMES IN THE CYCLOADDITION OF ACID HALIDES OR EQUIVALENTS TO CINNAMYLIDENEAMINES: A CONCISE NEW APPROACH TO 4-ACETOXYAZETIDIN-2-ONES¹.

J.M. Aizpurua, F.P. Cossío, B. Lecea, C. Palomo*

Departamento de Química Orgánica . Facultad de Ciencias Químicas. Universidad del País Vasco. Alza,San Sebastián . Spain.

Abstract: The reaction between 2-methylcinnamylideneamines and acid halides or equivalents leads to the stereospecific formation of cis- β -lactams. A new three steps approach to 4-acetoxyazetidin-2-ones as building blocks of β -lactam antibiotics is also described.

Although many methods are now available for the synthesis of β -lactams, the acid halide imine method or equivalent provided the most easy access to the β -lactam ring. However, from this method, it is often difficult to predict the stereochemistry of the product. For example, phthalimidoacetic acid generally gives <u>trans- β -lactams²</u> and alkoxyacetic acids often produce mixtures of <u>cis</u> and <u>trans</u> isomers, depending upon the experimental conditions employed³. Recently, we^{4a} and others^{4b} have found that the <u>cis</u> stereochemistry in the β -lactam formation increases when the bulk of the N-substituent is increased. Also, it has been described⁵ that cinnamylidene Schiff bases react with acid halides or anhydrides to afford azetidinones, generally with a high degree of cis stereoselectivity. These results prompted us to report our own, related



observations on the stereochemistry of β -lactam formation. Our finding is that increasing the bulk of the aldehyde in the starting Schiff base <u>2</u> the <u>cis</u> stereochemistry in the resulting β -lactam <u>3</u> predominates. Thus, we observed that methoxyacetyl chloride and phthalimidoacetyl bromide upon treatment with Schiff bases derived from cinnamaldehyde and 4-anisidine or 4-aminophenol yielded mixtures of <u>cis</u> and <u>trans</u> isomers,

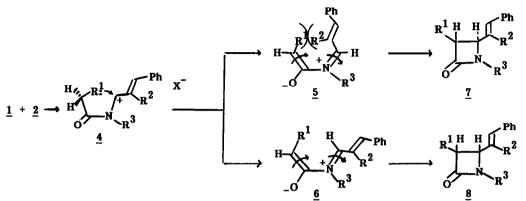
while 2-methylcinnamylideneamines, upon treatment with these acid halides yielded only the corresponding cis isomers. For example, whereas the Schiff base derived from cinnamaldehyde and 4-trimethylsiloxy-aniline gave an equimolar mixture of the cis and trans 3-phthalimido- β -lactam (entry 3), the same acid bromide afforded only the cis isomer when the Schiff base derived from 2-methylcinnamaldehyde and 4-trimethylsiloxyaniline was involved in the cycloaddition process (entry 4). Results of some experiments are given in Table 1 to illustrate the stereospecificity of the reaction, even when the reaction conditions favoured the trans stereochemistry^{3b}.

Entry	Product 3		Configuration ^a	
	\mathbf{R}^{1}	\mathbf{R}^2 \mathbf{R}^3	$\underline{cis}: \underline{trans}^{b}$ $\underline{cis}: \underline{trans}^{c}$	m.p.(ºC) ^d
1	Phth	H 4-CH ₃ OC ₆ H ₄	95 : 5 (45) 90 : 10	178-179
2	Phth	CH ₃ 4-CH ₃ OC ₆ H ₄	100 : 0 (64)	105-107
3	Phth	$H 4-Me_3SiOC_6H_4$	50:50	
4	Phth	CH ₃ 4-Me ₃ SiOC ₆ H ₄	$ 100 : 0 (50)^{I}$	172-174
5	MeO	H 4-CH3OC6H4	90: 10(40)	146-149
6	MeO	CH ₃ 4-CH ₃ OC ₆ H ₄	100 : 0(60) 100 : 0(40)	117-118
7	PhO	H 4-CH3OC6H4	100 : 0 (45)	180-182
8	PhO	CH ₃ 4-CH ₃ OC ₆ H ₄	100 : 0 (70)	157-158
9	ε ^e	CH ₃ 4-CH ₃ OC ₆ H ₄	100: 0(50)	150-151

Table 1. \underline{cis} : <u>trans</u> distribution in β -lactams 3

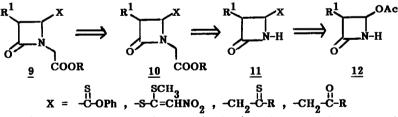
a) Cofiguration of C-3 and C-4 protons in all these β -lactams was determined by nmr spectroscopy. The numbers in parentheses indicate isolated yields of the β -lactam. b) The acid halide was added to a solution of the imine and triethylamine. c) Triethylamine was added to a solution of the acid halide and the imine. d) Melting point of the <u>cis</u> isomer. e) From Dane salt of aminoacetic acid and phenyl dichlorophosphate reagent, see ref. 10. f) The β -lactam was isolated with the free hydroxyl group.

Since, it is generally agreed that the acid chloride-imine reaction or equivalent proceeds through a dipolar intermediate of type 5 and 6, rather than by a concerted [2+2] addition⁵, a satisfactory explanation of these results could be the following one:

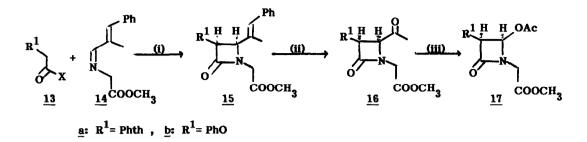


Abstraction of C_3 proton in the transition state <u>4</u> would give two zwitterions <u>5</u> and <u>6</u>. These zwitterions upon conrotatory ring closure lead to the corresponding <u>trans</u> and <u>cis</u> β -lactams <u>7</u> and <u>8</u> respectively. When cinnamaldehyde Schiff bases were involved in the cycloaddition process, intermediates <u>5</u> and <u>6</u> may be formed, as judged by the results obtained, however, when $R^2=CH_3$, zwitterion <u>6</u> is greatly favoured since the R^1/R^2 interaction is relieved and therefore the <u>cis-</u> β -lactam <u>8</u> results as the exclusive reaction product. Therefore, starting from carboxylic acids carrying heteroatoms with a pair of free electrons at the α -position and 2-methylcinnamylideneamines, now, seems to be possible to predict the stereochemistry of the resulting β -lactam⁶. Furthermore, it is worthy of note that these β -lactams <u>8</u> could serve as precursors of 4-acetoxyazetidin-2-ones <u>12</u> since the N-substituent on these β -lactams <u>8</u> can be N-dearylated by means of Kronenthal's method⁷.

These 4-acetoxyazetidin-2-ones $\underline{12}$ are known as useful building blocks of penem and carbapenem derivatives because they are easily convertible into various kinds of intermediates⁸. Recently, several research groups⁹ have described methodologies for obtaining 4-unsubstituted-1-alkoxycarbonylmethylazetidin-2-ones $\underline{10}$ as the last synthetic target for making $\underline{9}$.



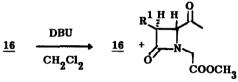
We have found that β -lactams of type <u>10</u> (X=OAc) can be prepared, stereospecifically, from β -lactams <u>15</u>, by a double oxidation sequence as shown in scheme 1. Thus, when the Schiff base <u>14</u>, derived from 2-methylcinnamaldehyde and methyl glycinate, was allowed to react with phthalimidoacetic acid <u>13a</u>, phenyl dichlorophosphate reagent and triethylamine, the <u>cis-</u> β -lactam <u>15a</u> was obtained in 66% yield as single isomer, as expected by the above observations. Oxidation of <u>15a</u> was achieved by means of potassium permanganate in the presence of NaH₂PO₄ in aqueous acetone-dioxane at room temperature to give <u>16a</u> in 80% yield. Using the same procedure, but starting from <u>15b</u> as crude product, the yield of <u>16b</u> decreased to 34%; however, ozonolysis of the crude compound <u>15b</u> in dichloromethane as solvent at -70°C afforded the desired β -lactam <u>16b</u> in 55% overall yield. We also have found that the corresponding <u>trans</u> isomers can be obtained by isomerization of these 4-acyl- β -lactams with 1,8-diazabicyclo (5.4.0) undec-7-ene (DBU) as catalyst. Thus, <u>16b</u> upon treatment with DBU for 60 min at room temperature yielded a mixture of <u>cis</u> and <u>trans</u> isomers in a molar ratio 65:35; when the reaction was allowed to stand at room temperature overnight an equimolar mixture of <u>16b</u> and <u>18b</u> was



Scheme 1. (i) PhOPOCl₂, CH₂Cl₂, NEt₃, r.t. 20-24h; (ii) KMnO₄, NaH₂PO₄, H₂O-Me₂CO, r.t. or O₃, CH₂Cl₂, -70^{\circ}C, then Me₂S or Pyr; (iii) MCPBA, Bz, 80^{\circ}C.

obtained. Interesting, <u>16a</u> upon treatment with DBU for 24h yielded a mixture of <u>cis</u> and <u>trans</u> isomers in a molar ratio 10/90. Under these conditions no isomerization was observed when <u>15a</u> was subjected to treatment with DBU.

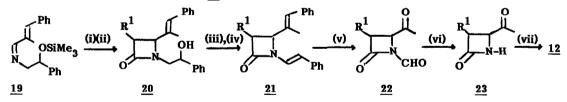
,				
<u>16</u>	time	Product <u>16</u>	distribution ^a <u>18</u>	(%)
a	24h	10	90	
Þ	1h 24h	65 50	35 50	



a) Determined by mmr spectroscopy

Finally, Baeyer-Villiger oxidation on these β -lactams <u>16</u> was carried out in refluxing benzene for 4-6hr by means of m-chloroperbenzoic acid (MCPBA) to give, after work-up, the expected 4-acetoxy- β -lactams 17.

Next, we extended our study to the synthesis of N-unsubstituted 4-acetoxyazetidin-2-ones <u>12</u>. For this purpose, we selected as starting materials the β -lactams <u>20</u> prepared by reaction between the corresponding acetic acid and the Schiff base <u>19</u> promoted by phenyl dichlorophosphate reagent. As was expected, these β -lactams also showed <u>cis</u> stereochemistry at C₃-C₄ protons. These β -lactams <u>20</u> upon treatment¹¹ with triphenylphosphine dibromide, followed by dehydrobromination by means of DBU in benzene yielded the corresponding N-styryl- β -lactams <u>21</u>.



Scheme 2. (i) RCH₂COOH, NEt₃, PhPPOCl₂, r.t. 20-24h; (ii) 1N HCl, Me₂CO; (iii) Ph₃PBr₂, CH₂Cl₂; (iv) DBU, Bz, reflux, 15min; (v) O₃, CH₂Cl₂, -70^{\circ}C; (vi) MeOH, NEt₃; (vii) MCPBA, CH₂Cl₂, r.t.

Simultaneous ozonolysis of the two double bonds on these β -lactams afforded the expected 4-acetyl--N-formylazetidin-2-ones <u>22</u>, (<u>22a</u> in 98% yield and <u>22b</u> in 80% yield). N-Deformylation was carried out under mild conditions by a simple modification of the known method¹². Thus treatment of <u>22a</u> in methanol with a catalytic amount of triethylamine yielded <u>23a</u> in 85% yield; using sodium methoxide in place of triethylamine only a 50% yield of <u>23a</u> was obtained. Isomerization of <u>23a</u> by means of DBU in dichloromethane as solvent for 24h at room temperature, gave the <u>trans</u> isomer in a molar ratio 10:90 (cis : trans). When <u>22b</u> was subjected to N-deformylation with methanol and triethylamine as catalyst, compound <u>24¹³</u> m.p: 118-120°C was isolated: ¹H-NMR (CDCl₃) δ ppm: 10.25 (m,1H,NH), 8.35(d,J=4Hz,1H,OCHO), 7.5-6.8 (m,5H,arom.), 5.3(s_b,2H,CH,CH), 3.65(s,3H,OCH₃), 2.25(s,3H,CH₃); IR(nujol) v cm⁻¹: 3250 (NH), 1750, 1720 (C=O), along with the corresponding β -lactam <u>23b</u>, m.p. 149-149.5°C, ¹H-NMR (CDCl₃) δ ppm: 7.4-7.0 (m,5H,arom.), 6.6(s_b,1H,NH), 5.5(d,d, J=5Hz, J'=2Hz, 1H,CH), 4.55(d, J=5Hz, 1H, CH), 1.9(s,3H,CH₃). Since β -lactams 23 can be easily converted into the corresponding 4-acetoxyderivatives under Baeyer-Villiger conditions^{3d,e} our procedure constitutes an atractive easy route to intermediates of β -lactam antibiotics. It is worthy of note that from this approach optically active compounds could be obtained from optically active Schiff bases of type 19.

In conclusion, we have found easy alternatives for the synthesis of a wide variety of building blocks of bicyclic β -lactam compounds. Total synthesis of β -lactam antibiotics starting from these compounds¹⁴ would be presented in the near future, as well as the scope of our method for producing <u>cis</u>- β -lactams starting from Schiff bases derived from 2-methylcinnamaldehyde and related aldehydes.

REFERENCES AND NOTES

- 1.- Contribution nº 60 in the series Reagents and Synthetic Methods. For the contribution nº 59 see J.M.Aizpurua, C. Palomo An.Quim. in press. The present work has been supported by Comision
 Asesora de Investigación Científica y Técnica (Project 994/84). A grant from Ministerio de Educación y Ciencia to F.P. Cossío is gratefully acknowledged.
- a) A.Arrieta, J.M. Aizpurua, C. Palomo, <u>Synth.Commun.</u>, <u>12</u>, 967 (1982); b) A.Arrieta, J.M.Aizpurua, C.Palomo, Tetrahedron Lett., 25, 3365 (1984).
- a) A.K.Bose, G.Spiegelman, M.S.Manhas, <u>Tetrahedron Lett</u>, 3167 (1971); b) A.K. Bose, Y.H.Chiang,
 M.S. Manhas, <u>Tetrahedron Lett</u>, 4091 (1972); c) A.K. Bose, B. Dayal, H.P.S. Chawla, M.S.Manhas,
 <u>Tetrahedron Lett</u>, 2823 (1972); d) A.K. Bose, J.C. Kapur, S.D.Sharma, M.S. Manhas, <u>Tetrahedron Lett</u>, 2319 (1973); e) A.K. Bose, B. Anjaneyulu, S.K. Battacharya, M.S.Manhas, Tetrahedron, 23,4769(1967).
- a) A. Arrieta, B.Lecea, C.Palomo, J.Chem.Soc.Perkin I, in press; b) H.W.Moore, G.Hughes, K. Srinivasachar, M.Fernandez, N.V. Nguyen, D. Schoon, A. Tranne, J.Org.Chem., 50, 4231(1985).
- 5.- T.W.Doyle, B. Belleau, B. Luh, C.F. Ferrari, M.P. Cunningham, Can.J.Chem., 55, 468 (1977).
- 6.- At this stage, we also have found that reaction between thiophenoxyacetic acid and imines <u>2</u> leads to the formation of the corresponding <u>trans</u> β-lactams. Such stereospecificity is not clear at present; see also, M.S. Manhas, S.S. Bari, B.M. Bhawal, A.K. Bose, <u>Tetrahedron Lett</u>, <u>25</u>, 4733 (1984).
- 7.- D.R. Kronenthal, C.Y. Han, M.K. Taylor, J.Org.Chem., 47, 2765 (1982).
- a) For a review, see: T. Kametani, <u>Heterocycles</u>, <u>17</u>, 463 (1982); b) Y. Ueda, S.C. Maynard, <u>Tetrahedron</u> Lett, 26, 6309 (1985); c) D.H. Hua, A. Verma, <u>Tetrahedron Lett</u>, <u>26</u>, 547 (1985).
- a) W.J. Leanza, F.DiNino, D.A. Muthard, R.R. Wilkening, K.J. Wildonger, R.W.Ratcliffe, B.G. Christensen, <u>Tetrahedron</u>, <u>39</u>, 2505 (1983); b) G. Johnson, P.M.Rees, B.C. Ross, <u>J.Chem.Soc.Chem.</u> <u>Commun</u>, 970 (1984); c) M.Hatanaka, H.Nitta, T.Ishimaru, <u>Tetrahedron Lett</u>, 2387(1984); d) M. Hatanaka, T.Hishimaru, <u>Tetrahedron Lett</u>, <u>24</u>, 4837 (1983); e) S. Hanessian, A. Bedeschi, C.Battistini, N.Mongelli, <u>J.Am.Chem.Soc.</u>, <u>107</u>, 1438 (1985).
- 10.- A. Arrieta, F.P. Cossío, C. Palomo, Tetrahedron Lett, 41, 1703 (1985).
- 11.- F.P. Cossío, C. Palomo, Tetrahedron Lett, 26, 4235 (1985).
- 12.- D. Häbich, Angew.Chem.Int.Ed.Engl., 22, 711 (1985).
- 13.- An analogous structure has been proposed when a β -lactam is formed by direct cyclization of β --amino acids, see: J.C. Sheehan, E.J. Corey, Org.Reactions, 9, 388 (1957).
- 14.- All the synthetic β-lactams are racemic mixtures. All new compounds had satisfactory microanalytical data and their proton magnetic resonance spectra (CDCl₃,ppm) fitted the assigned structures. Selected physical data includes: <u>15a</u> m.p. 191-194°C(EtOH), § 7.80(m,4H,arom.), 7.21(m,5H,arom.), 6.53 (m,1H,CH=), 5.55(d,J=6Hz,1H,CH), 4.78(d,J=6H,1H,CH), 4.35(d,d,J=20Hz,CH₂), 3.74(s,3H,CH₃), 1.60 (m,3H,=C-CH₃). <u>16a</u> m.p. 210-213°C (EtOH), §: 7.98(m,4H,arom.), 5.80(d,J=7Hz,1H,CH), 4.85(d, J=7Hz,1H,CH), 4.43(d,d,2H,J=18Hz,CH₂), 3.81(s,3H,CH₃), 2.16(s,3H,CH₃); <u>17a</u> m.p. 145°C (EtOH), §: 7.92(m,4H,arom.), 6.35(d,J=4Hz)1H,CH), 5.76(d,J=4Hz,1H,CH), 4.24(d,d,J=18Hz,2 H,CH₂), 3.82 (s,3H,CH₃), 1.97(s,3H,CH₃); <u>16b</u> m.p. 55-58°C (CCl₄), §: 7.28(m,5H,arom.), 5.55(d,J=6Hz,1H,CH), 4.86(d,J=6Hz,1H,CH), 4.28(d,d,J=18Hz,2H,CH₂), 3.72(s,3H,CH₃), 2.20(s,3H,CH₃); <u>17b</u> m.p: 83.5°C (CCl₄), §: 7.18(m,5H,arom.), 6.40(d,J=4Hz,1H,CH), 5.42(d,J=4Hz,1H,CH), 4.10(d,d,J=18Hz,2H,CH₂), 3.73(s,3H,CH₃), 2.00(s,3H,CH₃), <u>22a</u> m.p: 186-190°C, §: 8.9(s,1H,CHO), 7.8(s_b,4H,arom.), 5.8(d,J=7Hz,1H,CH), 4.95(d,J=6Hz,1H,CH), 2.0(s,3H,CH₃); <u>22b</u> m.p: 110-112°C, §: 8.95(s,1H,CHO), 7.5-6.8(m,5H, arom.), 5.6(d,J=6Hz,1H,CH), 4.95(d,J=6Hz,1H,CH), 2.95(s,3H,CH₂).

(Received in UK 23 June 1986)